Amendments to the Claims:

Cancel claim 2, without prejudice.

This listing of claims will replace all prior visions and listings of claims in the application:

List of Claims:

1. (currently amended) A method of applying a metal coating to graphite, comprising the steps of:

anodic etching said graphite in an alkaline etchant[[,]];

Pd seeding said graphite; and then

electroplating said graphite with said metal.

- 2. (canceled)
- 3. (currently amended) The method as set forth in of claim 2 1, further comprising the following step between said Pd seeding and said electroplating:

electroless plating said graphite to reinforce said Pd coating.

- 4. (currently amended) The method as set forth in of claim 3, wherein at least Ni or Cu is deposited in said electroless plating step.
- 5. (currently amended) The method as set forth in of claim 1, further comprising the following step between said anodic etching and a subsequent step:

directly transferring said graphite, obtained with said anodic etching step, into water or a weak aqueous solution.

- 6. (currently amended) The method as set forth in of claim 5, wherein between said anodic etching and said electroplating no ultrasound treatment is implemented.
- 7. (currently amended) The method as set forth in of claim 1, wherein said electroplating involves at least one of the following group metals: Ag, Cu, Ni and Sn.
- 8. (currently amended) The method as set forth in of claim 1, wherein said electroplating utilizes a current density in the range of 0.1 to 10 A/dm².
- 9. (currently amended) The method as set forth in of claim 1, wherein [[the]] a current duration in said electroplating is in the range of 5 to 90 minutes.
- 10. (currently amended) The method as set forth in of claim 1, wherein said anodic etching is done in alkaline etchant is a solution of at least one of NaOH and/or and KOH having a concentration in the range 10 to 70% by weight.
- 11. (currently amended) The method as set forth in of claim 10, wherein said anodic etching is done at a temperature in the range of 20°C to 70°C.

- 12. (currently amended) The method as set forth in of claim 1, wherein said graphite comprises graphite particles bound by plastics.
- 13. (currently amended) A method of fabricating a solder connection to a graphite component, comprising the steps of:

electroplating wherein, by [[a]] said method as set forth in of claim 1, [[a]] said metal coating is deposited on said graphite component, after which; and

applying a solder pad is applied to said metal coating as thus produced.

- 14. (currently amended) A <u>The</u> method as set forth in of claim 1, wherein said anodic etching is performed with an applied electrical potential in the range of 4V to 20V.
- 15. (currently amended) A <u>The</u> method as set forth in of claim 14, wherein said anodic etching has a duration in the range of 5 to 90 minutes, with the actual duration being inversely proportional to the applied electrical potential.
- 16. (new) A method of applying a metal coating to graphite, comprising the steps of: anodic etching said graphite in a solution of at least one of NaOH and KOH having a concentration in the range of 10 to 70% by weight; and then electroplating said graphite with said metal.
- 17. (new) A method of applying a metal coating to graphite, said graphite comprising graphite particles bound by plastics, the method comprising the steps of:

anodic etching said graphite in an alkaline etchant; and then electroplating said graphite with said metal.

18. (new) A method of fabricating a solder connection to a graphite component, comprising the steps of:

anodic etching said graphite component in an alkaline etchant; electroplating said graphite component with a metal coating; and then applying a solder pad to said metal coating as thus produced.

19. (new) A method of applying a metal coating to graphite, comprising the steps of: anodic etching said graphite in an alkaline etchant with an applied electrical potential in the range of 4V to 20V and a duration in the range of 5 to 90 minutes, with the actual duration being inversely proportional to the applied electrical potential; and then electroplating said graphite with said metal.